THE ISOXAZOLE AROMATIZATION REACTION. A SYNTHETIC APPROACH TO DITERPENOID PHENOLS

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During the investigation of isoxazole annelation^{1,2)} it was discovered that the quaternary salts of 4(3-oxoalkyl)isoxazoles were converted into acylphenols upon base treatment²⁾. This finding has drawn our attention to a new general synthetic approach to diterpenoid phenols. However, the reaction turned out to be fairly sensitive to reaction conditions, reflecting the character of isoxazoles. For instance, treatment of $I^{2)}$ with pyrrolidine in benzene afforded 3-acetyl-2-N-ethylamino-5,6,7,8-tetrahydronaphthalene(II), λ_{max}^{EtOH} 238, 267, 395 mµ(log ϵ 4.26, 3.83 and 3.65 respectively), V neat 3300, 1645, 1575, 1515 cm⁻¹, δ in CDCl₃ 1.28(3H, t. J=7 cps) 2.52(3H, s.) 3.22(2H, q. J=7 cps) 6.46(1H, s.) and 7.46(1H, s.) ppm, semicarbazone; m.p. 201-202°, in 25 % yield as a sole aromatic product, while with 1N sodium hydroxide an acetylphenol III^{2,4,8} became the major product(40-50 %) along with a small amount of II.



Similarly, treatment of the tetralone derivative IV, m.p. 104-105°, prepared in 60 % yield by MMC^{5} alkylation of tetralone with 3,5-dimethyl-4-chloromethylisoxazole³, with methyl iodide followed by reflux in 1N sodium hydroxide furnished a mixture of dihydrophenanthrene derivatives in about 40 % yield which consisted of V^{8} , m.p. 119-120°, λ_{max}^{EtOH} 222, 300 and 354 mµ (log ε 4.35, 4.30 and 3.81 respectively), Vnujol 1645 and 1625 cm⁻¹, δ in CDCl₃ 2.68(3H, s.) 2.90(4H, s.) 7.39(1H, s.) 7.60(1H, s.) and 12.18(1H, s.) ppm and VI, m.p. 115-117°, λ_{max}^{EtOH} 238, 262, 308 and 416(log ε 4.23, 4.32, 4.29 and 3.84 respectively), Vnujol 3300, 1640 and 1560 cm⁻¹, in about 2:1 ratio.



These reactions suggest that there are two ways of aromatization originating from a loss of a proton from either one of active methyl groups present at 3 and 5 positions of isoxazole as shown in the scheme.

Aside from the mechanistic interest, we were encouraged to learn that quaternary salts of 3,5-dimethyl-4(3-oxoalkyl)isoxazoles were transformed into acylphenols in moderate yields and tried to apply this reaction to a synthesis of ferruginol(XI), a typical diterpenoid phenol⁶⁾. Thus 5,5,8a-trimethyl-1decalone(VII)⁷⁾ was converted in the usual way into a β -ketoester derivative of which alkylation with 3,5-dimethyl-4-chloromethylisoxazole followed by decarbethoxylation with acid afforded VIII, m.p. 99-100°, \sum_{max}^{EtOH} 224 mµ(log & 3.67), \bigcup nujol 1704, 1638 cm⁻¹, δ in CCl₄ 0.94(6H, s.) 1.11(3H, s.) 2.12(3H, s.) 2.30(3H, s.) ppm. Treatment of VIII with triethyloxonium fluoroborate followed by reflux in 1N sodium hydroxide gave, along with some recovered starting material, an acylphenol IX⁸⁾, m.p. 123-124°, \sum_{max}^{EtOH} 220, 265 and 341 mµ(log & 4.33, 4.15 and 3.66 respectively), \bigcup nujol 1645 and 1621 cm⁻¹, δ in CDCl₃ 0.99(6H, s.)

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1.21(3H, s.) 2.60(3H, s.) 6.84(1H, s.) 7.35(1H, s.) and 11.87(1H, s.) ppm,in 36 % yield based on utilized VIII. The acylphenol IX was further characterized by the formation of its methyl ether (X), $\lambda_{\max}^{\text{EtOH}}$ 219, 258 and 321 mµ(log ϵ 4.12, 3.92 and 3.55 respectively), V chf. 1668 and 1608 cm⁻¹, δ in CDCl₃ 3.87 (3H, s.), semicarbazone; m.p. 218-219°¹⁰. Since X had been already converted into dl-ferruginol by King and his co-workers⁹) the isoxazole aromatization mentioned above thus completed a total synthesis of ferruginol¹¹.

Further applications of the method to syntheses of various diterpenoid phenols are in progress.





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- 1) G. Stork, Pure and Appl. Chem., 9, 931(1964)
- 2) G. Stork, S. Danishefsky and M. Ohashi, J. Am. Chem. Soc., in press
- 3) N. K. Kochetkov, E. D. Khomutova and M. V. Bazilevsky, <u>J. Gen. Chem.(USSR)</u>, <u>28</u>, 2376(1958)
- 4) M. P. O'Farrel, D. M. S. Wheeler, M. M. Wheeler and T. S. Wheeler, <u>J. Chem.</u> <u>Soc.</u>, <u>1955</u>, 3986
- 5) H. L. Finkbeiner and M. Stiles, <u>J. Am. Chem. Soc.</u>, <u>85</u>, 616(1963)
- 6) W. P. Campell and D. Todd, *ibid.*, <u>62</u>, 1287(1940)
- 7) a) J. D. Cocker and T. G. Halsall, <u>J. Chem. Soc.</u>, <u>1957</u>, 3441
 - b) F. Sondheimer and D. Elad, <u>J. Am. Chem. Soc.</u>, <u>79</u>, 5542(1957)
 - c) N. Ototani, T. Kato and Y. Kitahara, <u>Bull. Chem. Soc. Japan</u>, <u>40</u>, 1730(1967)
- 8) With ferric chloride this compound gives deep blue coloration.

9) F. E. King, T. J. King and J. G. Toplis, <u>J. Chem. Soc.</u>, <u>1957</u>, 573

- 10) Although melting point of the semicarbazone is slightly lower than the reported value(222-223°), its spectral properties are all in accord with structure X indubitablly.
- 11) All new crystalline compounds gave satisfactory elemental analysis.